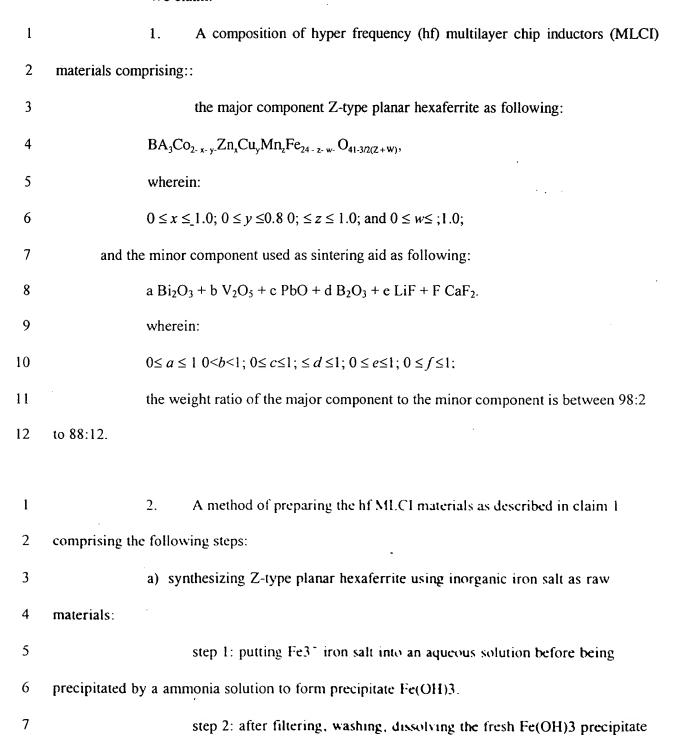
CLAIMS

We claim:



8	into hot citric acid solution at 60-800C with Fe/citric acid mole ratio in 1 to 2, obtaining a
9	transparent solution;
10	step 3: putting cobalt, barium, zinc, copper and manganese acetate or
11	nitrate salts in stoichiometric quantities to said solution in step 2, and then dropping an
12	appropriate ammonia until the said solution being neutral or slightly alkaline (pH'6-8) for 2
13	hours to give a stable sol containing the required Co2Z type hexaferrite composition;
14	step 4: drying said sol at 130-150 "C for 6— 1 Oh, and then heating-
15	treated at a temperature between 900 to 1250"C for 6h, resulting in Z-type hexaferrite
16	powders;
17	b) the sintering aids being mixed with hexaferrite powders by conventional
18	ceramic route:
19	step 5: mixing the said hexaferrite powders with sintering aids oxides
20	Bi2O3 V205 in a ball mill for 4 hours according the composition mentioned above during a
21	medium of water or alcohol to form a slurry;
22	step 6: drying said slurry at 80- 120 "C, then sieving the powders,
23	pressing them into pellets;
24	step 7: sintering said pellets at 870—-950C for 2-6h, obtaining the said
25	hyper frequency MLCI materials.
1	3 A method of preparing the hfMLCI materials as described in claim I
2	comprising the following steps:
3	a) synthesizing Z-type planar hexaferrite using organic iron salt such as iron
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4	citrate as raw materials:
5	step 1: dissolving iron citrate into aqueous solution, before mixing with
6	barium, cobalt, zinc, copper and manganese acetate or nitrate salts in stoichiometric quantities
7	to get a mixed solution;
8	step 2: dropping appropriate amount of ammonia solution into the said
9	mixed solution to make it neutral or slightly alkaline (pH=6-8), obtaining a steady sol;
10	step 3: drying the said sol at 130 to 150 "C for 6 to 1 Oh, then heating
11	treated between 900-—1250"C for 6h, resulting in Z-type hexaferrite powders;
12	b) the sintering aids being mixed with hexaferrite powders by conventional
13	ceramic route:
14	step 4: mixing the said hexaferrite powders with sintering aids oxides
15	Bi2O3 V2O5 in a ball mill for 4 hours according the composition mentioned above during a
16	medium of water or alcohol to form a slurry;
17	step 5: drying said slurry at 80—I 20"C, then sieving the powders,
18	pressing them into pellets;
19	step 6: sintering said pellets at 870—~950°C for 2-6h, obtaining the said
20	hyper frequency MLCI materials.
1	4. A method of preparing the hf MLCI materials as described in claim 1
2	comprising the following steps:
3	a) synthesizing Z-type planar hexaferrite using inorganic iron salt as raw
4	materials:

5	step 1: putting Fe ³⁺ iron salt into an aqueous solution before being
6	precipitated by a ammonia solution to form precipitate Fe(OH)3.
7	step 2: after filtering, washing, dissolving the fresh Fe(OH)3 precipitate
8	into hot citric acid solution at 6080"C with Fe/citric acid mole ratio in 1 to 2, obtaining a
9	transparent solution;
10	step 3: putting cobalt, barium, zinc, copper and manganese acetate or
11	nitrate salts in stoichiometric quantities to said solution in step 2, and then dropping an
12	appropriate ammonia until the said solution being neutral or slightly alkaline (pH6-8) for 2
13	hours to give a stable sol containing the required Co2Z type hexaferrite composition;
14	step 4: drying said sol at 130-150"C for 6~10h, and then heating-treated
15	at a temperature between 900 to 1250"C for 6h, resulting in Z-type hexaferrite powders;
16	b) the sintering aids being added into hexaferrite powders by chemical coating
17	route as the following:
18	step 5: dispersing the said hexaferrite powders into ethylene glycol to
19	form a slurry, blending the sintering aids in water-soluble forms into the slurry, adjusting pH
20	value of the mixed slurry so as to the sintering aids coating on the surface of hexaferrite
21	particles in forms of hydroxides;
22	step 6: after drying the mixed slurry, calcining it at 700°C for 2h to
23	form a second hexaferrite powders containing sintering aids;
24	step 7: sieving, pressing the second powders and sintering them at
25	870~950C for 6h, obtaining the invented hyper frequency MLCI materials.

1	5. A method of preparing the hf MLCI materials as described in claim 1
2	comprising the following steps:
3	a) synthesizing Z-type planar hexaferrite using organic iron salt such as iron
4	citrate as raw materials:
5	step 1: dissolving iron citrate into aqueous solution, before mixing with
6	barium, cobalt, zinc, copper and manganese acetate or nitrate salts in stoichiometric quantities
7	to get a mixed solution;
8	step 2: dropping appropriate amount of ammonia solution into the said
9	mixed solution for 6 to 8 hours to make it neutral or slightly alkaline (pH=6-8), obtaining a
10	steady sol;
11	step 3: drying the said sol at 130 to 150 "C for 6 to 1 Oh, then heating
12	treated between 900~1250"C for 6h, resulting in Z-type hexaferrite powders;
13	b) the sintering aids being added into hexaferrite powders by chemical coating
14	route as the following:
15	step 4: dispersing the said hexaferrite powders into ethylene glycol to
16	form a slurry, blending the sintering aids in water-soluble forms into the slurry, adjusting pH
17	value of the mixed slurry so as to the sintering aids coating on the surface of hexaferrite
18	particles in forms of hydroxides;
19	
20	step 5: after drying the mixed slurry, calcining it at 700 "C for 2h to form
21	a second hexaferrite powders containing sintering aids;

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- 22 s step 6: sieving, pressing the second powders and sintering them at
- 23 870~950"C for 6h, obtaining the invented hyper frequency MLCI materials.